



FIG. 2. Isobaric densities of liquid  $\text{He}^4$  as a function of temperature for pressures up to 4.5 atmospheres.

value by more than 2%. We believe that the resulting uncertainty in  $k_T$  is about  $\pm 3\%$ , except for one point at the SVP at  $5.000^\circ\text{K}$  where the uncertainty is about 15%. Figure 3 shows these isothermal compressibility results as a function of pressure along the five isotherms. Keesom and Keesom (Keesom 1942, p. 243) obtained a smoothed graph of  $(\partial\rho/\partial P)_T$  against  $T$  from which the isothermal compressibility of liquid helium may be obtained at 2.5 atmospheres and 3.0, 3.5, and  $4.0^\circ\text{K}$  to compare with the present measurements. After correction to the 1958 scale of temperatures, Keesom and Keesom's values are between 2 and 5% higher than our present results.

Table I also shows values of the ratio of heat capacities  $\gamma$  calculated from equation (3.4). Atkins and Stasior's (1953) smoothed values of  $u_1$  were used, interpolating graphically where necessary. If these  $u_1$  values are uncertain by  $\pm 2\%$ , then our  $\gamma$  values are uncertain by  $\pm 5\%$ . As no measurements have been made of  $u_1$  at  $4.500^\circ\text{K}$  or  $5.000^\circ\text{K}$ , no calculation of  $\gamma$  was possible for those two isotherms.

Finally, Table I shows values of the liquid structure factor  $\mathcal{L}_0$  calculated entirely from the results of the present measurements, using equation (3.5). The estimated uncertainty in  $\mathcal{L}_0$  is the same as that for  $k_T$ , namely  $\pm 3\%$ , except for the one point at the SVP at  $5.000^\circ\text{K}$  where the uncertainty is about 15%. Gordon, Shaw, and Daunt (1954) have measured scattering of X rays down to angles of  $1.5^\circ$  at  $4.2^\circ\text{K}$  at the SVP. Their data extrapolated to zero angle gives  $\mathcal{L}_0 = 0.575 \pm 0.040$ .<sup>\*</sup> Our results listed in Table I may be

<sup>\*</sup>Due to a misprint, their paper states 0.475, but their Fig. 2 shows that 0.575 is meant for this quantity.